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### *HYDROTHERMAL FLUIDS AND PETROLEUM IN SURFACE SEDIMENTS OF GUAYMAS BASIN. GULF OF CALIFORNIA: A CASE STUDY*

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#### *Abstract*

*Surface sediments in hydrothcrmally active zones of the Guaymas Basin, Gulf of California, have been analyzed for the inorganic chemical composition of sediments and their interstitial waters, as well as for their organic che mistry. Analysis of the inorganic constituents of vent fluids establishes end-member concentrations of hydrothermal fluids and also indicates that little change in chemical com position has occurred between 1981 and 1988. Special atten tion is given to short surface cores collected with DSV ALVIN in areas characterized by upward advection of fluids of hydrothermal origin, both in areas of strong advective flow as well as seepage through large bacterial mats of Beggiatoa. Interstitial-water compositions suggest that fluids consisting of hydrothermal and scawater components rise to the surface, often carrying a supply of sulfide of shal low origin; this in turn may serve as an energy source for the growth ofsulfide-oxidizing bacteria. Studies of the com position of hydrothermally generated petroleum show that it is a condensate (C|-C2o hydrocarbons and polynuclcar aromatic hydrocarbons), and much of this material has migrated with the upward-advccling fluids from short depths below the sediment-water interface. The petroleum in the sediments is found mainly in fractures in silicacemented clasts and it is biodegraded. In the vent/mound, the petroleum is an unaltered condensate, indicating active migration that exceeds biodegradation.*

*Keywords: Beggialoa mats, condensate, diffuse flow, gas. Gulf of California, Guaymas Basin, hydrothermal fluid composition, hydrothermal petroleum, interstitial-water composition, organic carbon, polynuclcar aromatic hydrocarbons, sulfide generation.*

#### *SOMMAIRE*

*Les sediments dc surface des zones a manifestations hydrothcrmalcs dans le bassin dc Guaymas, dans le golfe de Californic, ont etc analyses afin de determiner la com position chimiquc (fractions inorganique et organiquc) des sediments et dc I'cau intcrstiticlle. L'analysedc la fraction inorganique dc 1'eau des events definil la concentration des elements dans Ics fluides hydrothcrmaux avant leur dilu tion, et momrc que tres peu dc changements sonl survc-* *nusentrc 1981 et 1988. Nous avons porte une attention tome speciale aux courtes carottcs prelevecs dans la couche de surface au moyen du submersible DSV ALVIN; les prelevements ont die faits dans des regions d'advection verti cal des fluides d'originc hydrothcrmale, aussi bien dans des zones a fort ccoulcmcnt advectif que prcs des ecoulements diffus a travers Ics cpaisscs accumulations de la bactcric Beggialoa. La composition des caux interstiticlles montrc que les fluides qui contiennent une composante hydrothcrmale ainsi que I'cau de mer atteigncnt la surface, et sont gencralcment munis de sulfurc d'originc peu profonde; ce milieu peut done fournir une source d'energie pour assurer la croissance de bactcries qui catalyscnt l'oxydation dc sulfure. Nos etudes sur la composition du petrole d'originc hydrothcrmale montrem qu'il s'agit d'un condense (hydrocarburcs C|-C20 el hydrocarbures polynucleaires aromatiques); en grande partic, cettc fraction a migre'avec les fluides transported par advection verticale a partir d'une faible profondcur en dessous de I'interfacc cau-scdiment. Le petrole des sediments sc trouvc surtout dans les fissu res des fragments a ciment siliceux, et est transforme par biodegradation. Pres de I'event, le petrole est un condense non allcre, cc qui indiquc que le taux dc migration active dans cc milieu surpassc le taux de biodegradation.*

*(Traduit par la Redaction)*

*Mots-clis: accumulations dc Beggialoa, condense, ecoule mcnt diffus, gaz, golfe dc la Californic, bassin de Guay mas, composition du fluidc hydrothermal, petrole hydrothermal, composition de l'eau intcrstiticlle, carbone organique, hydrocarburcs polynucleaires aroma tiques, formation de sulfurcs.*

#### *Introduction*

*The hydrothermal system in Guaymas Basin ofthe Gulf of California (Fig. 1) is of special interest in the study of hydrothermal ore deposition, especially because here newly formed oceanic basement con sists either of sill intrusions in rapidly deposited, organic carbon-rich sediments, or of massive intru sions at greater depths (Moore 1973, Williams et al. 1979, Einsele et al. 1980, Gieskes et al. I982a,b, 1988). Hydrothermal interactions involve both the alteration of basaltic rocks and of sediments, the lat-*



*Fic. I. Map of study area - note cores 1617-AC1, 1619- AC2, 1629-AC2, 1630-ACI, and 1984-AC1, all in the southern hydrothermal field. Shaded areas represent extent of sill complexes (Lonsdale & Becker 1985).*

*ter also providing a medium for the thermal genera tion of hydrocarbons (Einsele et al. 1980, Kastner 1982, Simoneil 1983, 1984a,b, 1985a,b, Simoneit & Lonsdale 1982, Kawka & Simoneit 1987, Simoneit et al. 1984).*



*Fig. 2. Cartoon version of hydrothermal deposit - ALVIN dive 1630. A: diffuse flow, temperature <206°C; B: yellow bacterial mat; C: yellow bacterial mat, sampled for core 1630-ACI; D: orange bacterial mat; E: hydrothermal spires(1-1.5 m tall); F: fault zone (<30cm).*

*Studies of Deep Sea DrillingProject (DSDP) cores obtained from Sites 477, 478, and 481 involved interstitial-water chemistry (Gieskes et al. 1982a), solid-phase geochemistry and mineralogy (Kastner 1982, Kelts 1982, Niemitz 1982), and physical proper ties of the sediments (Einsele el al. 1980, Einsele 1982).The results led Kastner (1982) and Gieskes el al. (1982b) to conclude that two major hydrother mal systems could be distinguished in Guaymas Basin:*





*Fig. 3. Compositional changes versus magnesium concen trations in hydrothermal vents of southern hydrother mal field, Guaymas Basin, 1988.*

*(1) Hydrothermal activity associated with relatively thin, shallow basaltic sill intrusions in highly porous sediments. This activity is usually ofshort duration and is associated with temperatures <200°C in the vicinity of the sills. Decreased porosities indicate that expulsion of pore waters accompanies the sill emplacement.*

*(2) Hydrothermal activity associated with large magmatic intrusions at greaterdepths; this activity occurs in a deep-seated, recharging, circulation system. Hydrothermally altered waters, after reaction with underlying basalts, interact with overlying sediments and lead to greenschist formation at temperatures of >300°C (Kastner 1982, Kelts 1982). Mass-balance considerations of oxygen isotopes (Kastner 1982) indicated that open-system conditions prevailed dur ing greenschist formation.*

*Lonsdale & Becker (1985) suggested the presence of an additional hydrothermal system in which slow recharging convection occurs within the sediments overlying the sills. However, this type of system appears to be part of the first type described above.*

*Hydrothermal fluids have been observed to exit the sediment column in several ways (Lonsdale & Becker 1985). Fluids emanate from hydrothermal mounds and chimneys at temperatures up to 3I5°C (Von Damm et al. 1985). They seem to have their origin at greater depths, and must have flowed rapidly along fissures and faults in the sedimentary column. Hydrothermal deposits typically occur along fault zones in the sediments. Fluids also emanate diffusely through the sediments. Areas of diffuse flow are typically characterized by shimmering water just above the sediments, or by bacterial mats of giant Beggialoa on top of the sediments.*

*High-temperature vent fluids emanating from hydrothermal mounds and chimneys have been studied by Von Damm et al. (1985) and Campbell et al. (1988). Their chemical composition is similar to that of the fluids in the deeper parts of DSDP Hole 477A (Gieskes et al. 1982a). Together with the often extensive size of the hydrothermal deposits, this sug gests that these fluids are associated chiefly with deeper-seated systems, which should operate on a longer basis than sill-induced systems.*

*In the present study we investigated the chemis try of sediments and interstitial waters associated with diffuse flow through the sediments. Of impor tance is the question with regard to the origin of hydrothermal fluids in this type of system. Diffuse flow has been observed both in the vicinity of large hydrothermal deposits, and in zones removed from such deposits. It is of interest to determine whether fluids in this type of system are associated mainly with a sill-induced hydrothermal system, with a deeper-seated system, or with both.*

*During DSV ALVIN operations in Guaymas Basin in 1982 and 1985 (Gieskes et al. 1988), a series of*

*short push cores was obtained in zones of diffuse flow or below Beggialoa bacterial mats. In the present study we concentrate on one ALVIN dive area in particular, but discuss the results in a broader context.*

#### *Study Area*

*In August 1985 several dives were made in the southern vent field of the southern trough of Guaymas Basin (Fig. 1). In particular, core 1630- ACI was obtained in an area where hydrothermal deposits indicated only very recent growth along a*



	Campbell et al. (1988)	This work
$Cl^*(mM)$	597-617	610
$Mg^{2*}$ (mM)	0	0
$SO_4^{2*}$ (mM)	0	0
$Ca2+$ (mM)	28	27
NH <sup>+</sup> (mM)	15	13
H <sub>2</sub> S (mM)	6	6
$K^*$ (mM)	43	47
Mn (µM)	140	110

*TABLE 2. VISUAL OBSERVATIONS OF "ALVIN" CORE 1630-ACI*





*Fig. 4. Detail of "oil shale" piece. Core 1630-ACI, sec tion 7.*

fault zone, at least 30 m away from larger hydrother*mal mounds. A cartoon of the area is presented in Figure 2. One push core (1630-ACI) was obtained in a yellow Beggialoa bacterial mat, and another push core (1630-PC) sampled the top of one of the hydrothermal spires. The material of this hydrother mal spire consists mainly of barite and precipitated amorphous silica, saturated with petroleum conden sate. Upon rising through the thermocline, several chimney pieces jumped out of the collection basket as a result of the release of pressurized gas and oil, which rose ahead of the ALVIN, thus causing the submarine to surface in a large oil slick. Core 1630- PC, of about 25 cm length, was half-filled with chim ney material and the liquid phase above this was about one half water and one half oil. Below we dis cuss the nature of this oil compared to that found trapped between cemented pieces of silica in core 1630-ACI. Interstitial water samples were obtained from push core 1630-ACI, and these were analyzed for major ionic constituents (hydrogen sulfide onboard ship; major constituents in home labora tory). Solids analyses were carried out using wetchemical techniques (Donnelly 1980, Brown, M.S. thesis in prep.).*

*Also studied were core 1617-AC1, obtained from an area of strong diffusive flow, and cores 1619- AC2, 1629-AC2, and 1984-AC1 from areas of bac terial mats. Interstitial-water data obtained from these cores arc compared with results from core 1630- ACI and hydrothermal fluids obtained from hydrothermal vents in the southern vent field in 1988.*

#### *Results*

#### *Inorganic chemistry*

*Vent fluids. During February 1988 two of us (BRTS and AM) obtained hydrothermal fluids from*

*vents in the southern vent field (Fig. 1). End-member concentrations of hydrothermal fluids are obtained from extrapolation of correlation plots of concen trations* of various constituents *versus* those of mag*nesium (Von Damm et al. 1985,Campbell et al. 1988; Fig. 3). The common assumption is that the endmember fluids contain no Mg2+ or SO.,2-. Extrapo lated hydrothermal fluid concentrations are com paredwith those presented by Campbell el al. (1988) and, in agreement with the latter authors' observa tions, no significant change has occurred over the last seven years (Table 1). This attests to the longevity of the hydrothermal system associated with mound formation in the southern hydrothermal field.*

*Core 1630-ACI - solids. In Table 2 we present* visual observations of centrifuged materials obtained *during the extraction of the interstitial fluids from the sediments. The petroliferous nature of the sedi ments is most apparent below a depth of -8 cm, coinciding with the depth at which most cementa tion has occurred in this core (see below). Numer ous silica-cemented clastsoccur in the sediments, par ticularly in section 6 (7.5-9 cm). Figure 4 presents a closeup of one clast demonstrating the "reservoir rock" nature of these clasts, i.e., it appears to sand wich bitumen-like material. X-ray diffraction anal ysis of this clast revealed the presence of mostly amorphous silica with minor amounts of clay (SiO<sub>2</sub>) -89.1%; AI2Oj - 1.83%; Ca -1.94%). Scanning electron microscope (SEM) observations have rev ealed that the clasts consist mainly of cemented dia toms, the cement appearing to be amorphous silica. The cement may have formed by the precipitation of silica from dissolved silica-rich hydrothermal fluids. In part this silica enrichment of the fluids can have been the result of diatom dissolution at depth.*

*The bulk chemical composition of the sediments is presented in Figure 5. The concentration of Al is given in wt.% (salt-corrected), whereas the other ele ments are presented as mole ratios with respect to Al. Such mole ratios are often indicators of chemi cal change in the sediments. Ti/Al ratios, and espe cially Fc/AI ratios, are quite variable in the upper 10cm of the core. Si/Al ratios are high in the upper 12cm of the core representing a substantial biogenic silicacomponent; lower ratios at greaterdepths may signify dissolution of diatoms. The increase in K/AI below 10 cm is significant and probably represents uptake of K \* in clay minerals.*

*Core 1630-ACI - interstitial water. The depth dis tributions of all components (Fig. 6) with the excep tion of sulfide show a change of slope at a depth of -8 cm, i.e., in the zone where the largest amounts of clasts have been observed. Sulfide concentrations reach a maximum at -7 cm depth. We suggest that the silica cement in this layer is laterally continuous and that it represents a barrier to upward flow of hydrothermal fluids which, judging from the chem-*







*Fig. 7. Correlation between imcrslilial water components and magnesium in core 1630-ACI.*

*ical changes below this depth, appear to dominate the interstitial fluids. Silica cements might allow upward diffusion of dissolved constituents, but not actual fluid flow.*

*Of particular interest is the dissolved H2S profile, which shows a distinct maximum at a depth of ~7 cm. To investigate the problem of whether signifi cant reactions have affected the concentration-depth profiles within the coring interval, we have assumed Mg2+ to be conservative, i.e. the concentration changes represent mixing between seawater and a modified hydrothermal fluid. Figure 7 presents corre lation* plots of several constituents *versus* magnesium *concentrations. It is evident that for K+, Li+, Ca2\* and CI", conservative behavior can be inferred, but not for S042~ or, to some extent, NH4+. The Mn2\* concentration variations indicate production of Mn2\* within the sampling interval.*

*The sulfate trend implies removal of S042~, presumably as a result of sulfate reduction in the upper 8 cm. This would be consistent with rapid increase in sulfide in this interval (Fig. 6). In addi tion, a slight production of ammonia is evident in* *the upper 8 cm, presumably as a result of ammonification processes associated with the sulfate reduc tion process. It does appear, however, that the high dissolved NH4 \* and the low S042" at greater depths arc primarily due to a hydrothermal component.*

#### *Comparison to other ALVIN cores*

*Figure 8 illustrates interstitial water profiles for cores 1617-ACI, 1619-AC2, 1629-AC2, 1630-ACI and I984-AC1. All these cores come from dives in the southern vent field (Fig. 1). Core 1617-ACI was obtained in a zone characterized by strong upwelling of hydrothermal fluids (evident from flow through holes in the sediment); cores 1619-AC2, 1629-AC2 and 1984-AC1 were taken in bacterial mats. The concentration profiles of Ca2+ and Mg2\* suggest that the interstitial waters, although show ing typical upward curvature, usually associated with upward migration of fluids (Maris & Bender 1982, Sayles & Jenkins 1982), do appear to be character ized by mixtures of seawater and hydrothermal fluids in* different proportions. Estimates of the hydrother*mal contribution are based chiefly on the Mg2\* concentration, which should be vanishingly low in hydrothermal fluids (Table I). Cores 1617-ACI and 1630-ACI contain at least a 90% hydrothermal com ponent in their deeper parts, whereas core 1619-AC2 shows the least hydrothermal component.*

*Of interest are the profiles of dissolved sulfate. Though in core 1617-ACI changes in Mg2' and S042" are linearly correlated, this is not the case in cores 1629-AC2, 1630-ACI, 1984-AC1 and perhaps in 1619-AC2. This strong sulfate depletion can be understood in terms of sulfate reduction processes taking place in the upper 10 cm of the sediments, leading to the production of significant amounts of sulfides (Fig. 5). Upward transport of this sulfide would then provide the necessary sulfide to the bac terial mats for continued growth. Sulfate reduction processes arc probably enhanced as a result of higher prevailing temperatures in these sediments. Thus, it is likely that the added transport ofsulfide towards the surface of the sediments as a result of upward convection of highly diluted hydrothermal fluids cre ates the appropriate conditions for the growth of bac terial mats on top of the sediments in many areas of the Guaymas Basin.*

*Figure 9 is a correlation diagram for the various "end-member" compositions of the ALVIN cores, i.e., the compositions at which the concentrations of most of the constituents become constant with depth. The data are plotted versus changes in con centration from that in seawater using Mg2\* as a reference on the assumption that this constituent shows conservative behavior. For comparison the data of the lowermost part of DSDP Hole 477A (Gieskes et al. 1982a)* and those of the average vent*fluid composition of the southern hydrothermal field*



*Fig. 8. Interstitial waler chemistry in cores 1617-ACI, 1619-AC2, 1629-AC2, 1630-ACI, and 1984-ACI.*

*(Von Damm et al. 1985, Campbell el al. 1988;Table 1) are also plotted. Changes in Li\* are very similar to those expected from a mixture of seawater and of the end-member hydrothermal fluid. Though K\* concentration changes appear to follow a welldefined mixing line, the change in K\* calculated for the hydrothermal end-member*  $( \Delta K^+ \sim 23 \text{ mM} )$  *is much smaller than that observed in the vent fluids (AK \* - 37 mM). For changes of CI" concentration, the observations suggest higher concentrations than expected from vent fluids. For S042~, NH4\*, and Ca2\*, considerable scatter occurs, indicating that reactions in the sediments may strongly affect the data. This, especially for Ca2\*, is also evident from Figure 8, in which Ca2\* changes do not necessarily reflect changes in Mg2\*. Smaller changes in Ca2\* in some cores may be the result of CaS04 precipi tation at greater depth (Stout & Campbell 1983, Gieskes et al. 1988).*

*Two processes must be considered to explain differences between observed changes in composi tion and those expected from simple end-member mixing. The first possibility is that potential artifacts*

*in the data may occur as a result of the interstitialwater retrieval process. Core samples are extracted at room temperature (- 25°C), whereas in situ tem peratures* may have been as high as 100°C. For K<sup>\*</sup> *this could lead to lower K\* concentrations, i.e., artifacts would be opposite to those found in DSDP sediments (Sayles & Manhcim 1974, Gieskes 1974). On the other hand, K\* concentration changes of DSDP Site 477 arc in close agreement with those of vent fluids. The other process involves reactions in the sediments during relatively slow percolation of hydrothermal fluids through the sediments. Uptake of*  $K^+$  at *temperatures* below  $\sim 200^{\circ}$ C *into authigenic feldspars was noted in Site 477 (Kastner 1982), and solids data on core 1630-ACI (Fig. 5) also indi cate K\*uptake. Changes in S042" and NH4\* differ ent from those expected from simple mixing of hydrothermal end-member fluid and seawater can be understood in terms of enhanced bacterial activity. In addition, possible precipitation of CaS04 at depth (Stout & Campbell 1983, Gieskes et al. 1988), as well as possible dissolution of calcium carbonate (Von Damm et al. 1985) will affect con-*



*Fig. 9. Correlation plots for hydrothermal end-member composition of ALVIN cores. Hole 477A(Gieskes et al. 1982a),* and average vent fluids (o) in southern field (Von Damm et al. 1985, Campbell et al. 1988; Table 1). Data in terms *of changes from seawater concentrations.*



*Fig. 10. Total organic carbon in ALVIN cores 1617-ACI, 16I9-AC2,1629-AC2and 1630-ACI. Solid circles: 1617- ACI; open circles: I6I9-AC2; triangles: I629-AC2; crosses: 1630-ACI.*

*centration changes in Ca-\*. Similarly, higher-thanexpected changes in CI" concentration may be the result oi hydration reactions in the sediments.*

*The end-member hydrothermal fluid composition used in the above comparison is derived from vent fluids. Because of the constancy of this composition over a long period of time, we interpret the endmember to be a result of a deep-seated hydrother mal system (Kastner 1982, Gieskes et al. 1982b). Unfortunately, because of the possibility of modifi cations due to reactions during the slow ascent of the fluids through the sediment column, we are not able to judge whether the differences from the sim ple mixing line are due to a different hydrothermal end-member, i.e., associated with a sill-induced sys tem. However, there is little doubt that subsurface mixing of seawater and hydrothermal fluids does occur, and that local chemical reactions near the sur face (e.g., silica dissolution, gypsum precipitation, sulfate reduction) can act as important modifiers to the chemical composition of the interstitial waters. Below we show evidence also that important modifi cations of organic matter occur in the sediments when compared to organic materials obtained from hydrothermal vents.*

#### Organic Geochemistry

#### **Organic carbon**

*Determinations of organic carbon contents were carried out on dried, ground, bulk samples using a coulometric method. In all cores, organic carbon contents decrease with depth from surface values of* *about 4% organic C to 1.5-2.5% at 10 cm subbottom (Fig. 10). In core 1630-ACI, a distinct decrease occurs below the clast-rich layer. Organic matter diagenesis may have been more extensive in cores 1619-AC2and 1629-AC2, possibly because of a longer period of exposure to enhanced tempera ture conditions and hence a more advanced stage of organic carbon catagenesis. Temperature gradients in such sediments have been measured during ALVIN dives in 1988, and typical values in mat areas are: 5 cm -5°C; 10 cm -25°C, 15 cm -50°C, 50 cm -90CC, I m -110°C (Sayles unpub. data, Somoneit unpub. data).*

#### *Petroleum*

*The occurrence of petroleum in the sediments of Guaymas Basin is one of the chief characteristics of hydrothermal processes in this area (Simoneit 1983, 1984a,b, 1985a,b, Simoneit & Lonsdale 1982, Simoneit et al. 1984). The hydrothermal spire sam ple (1630-PC) contained abundant oil, and bitumen residues were found in clasts and mud of core 1630- ACI (Fig. 2). Lighter normal hydrocarbons from the latter had vanished mostly due to the effects of bio degradation, and to some extent due to recovery, storage, and drying. However, the remaining bitu men portion is of particular interest as it contains the heavier non-biodegraded remnants.*

*The total n-heptane extracts of mud (sediment) and clasts from core 1630-ACI were analyzed by gas chromatography (GC, operating conditions as in Simoneit 1984a) and the representative traces are shown in Figures 11 and 12, respectively. Sediment samples 6, 7 and 10 (Figs. 1lb.c.d.) are identical in that their extracts consist primarily of aromatic hydrocarbons, and the same is the case for clasts 6 and 7 (Figs. 12a,b,c,). Three samples from section 7 (namely sediment, "shale" 7, and another clast 7) are all identical (Fig. lie, Figs. 12b,c). "Shale" 7 was analyzed by gas chromatography - mass spec trometry (GC-MS) and details are given below. The clasts from sections 8 and 9 are, however, different. Their extracts are comprised primarily of low amounts of unidentified compounds. Also, the sedi ment from section <sup>2</sup> contains <sup>a</sup> different suite of compounds as discussed below.*

*Sample 1630-ACI,7 from one of the "oil shale" clasts in section 7 (Fig. 4) was analyzed by GC-MS. The total hydrocarbons are comprised mainly of unresolved branched and cyclic components (also termed unresolved complex mixture, UCM); the major resolved chromatographic peaks are poly nuclcar aromatic hydrocarbons (PAH), thio-PAH and alkyl-PAH (Fig. 13a). The major PAH consist of phenanthrene (Fig. 14a), anthracene, fluoranthene, pyrene, benzanthracene, chrysenc, benzfluoranthene, benzpyrenes and perylene. A hightemperature component (> —300°C) reflecting* *hydrothermal activity is indicated by this PAH dis tribution coupled with the presence of pericondensed alicyclic PAH (e.g., methylenephenanthrene. Fig. 14b; Simoneit 1983, I984a,b). Biomarkers used as petroleum maturity indicators, such as triterpancs, steranes and extended tricyclic terpanes, are not detectable in this sample. Their absence, cou pled with the resolved compound and UCM distri butions, support an origin from a biodegraded con densate. However, various maturity indices can be calculated for the dimethylnaphthalenes (DNR) and methylphenanthrenes (MPI) (Fig. 14; Radke & Welte 1983, Radke elal. 1982, Alexander etal. 1984, 1985). These indices (Table 3) are within the range typical of immature petroleums. The immaturity, coupled with the high thermal gradient and observed oil dis charges, are consistent with the interpretation of a short migration distance, for petroleum in this sam ple, from source sediment to the seabed (Kawka & Simoneit 1987); the immaturity also indicates that sediment samples 6, 7 and 10, and clast sample 6, can be interpreted analogously. Long distance (> 1 km) migration at low temperature would be expected to deplete the high molecular weight PAH in the petroleum.*

*The GC-MS data for the total extract of the sedi ment from section 2 (Fig. 1la) indicate it to be a mix ture of thermogenic and natural products. The dominant peak is phenanthrene, and the clusters are various series* of alkyl PAH with mainly  $C_2$  and  $C_3$ *alkyl substitution. These compounds are of a ther mogenic origin and probably represent the watersoluble fraction of the condensates encountered deeper in the core. The cluster of peaks at longer GC retention time consists of various natural product derivatives such as mainly hopenes with exocyclic double bonds, stanols and other unknown C28 and C30 triterpenoids. Normal and isoprenoidal alkanes are not detectable, and their absence compared to other shallow sediments from the basin (e.g., Simoneit et al. 1979) indicates that this sample has been biodegraded. The presence ofthe labile hopenes indicates that the sediment has not been heated > - 50°C, and thus the aromatic compounds prob ably have migrated as a cold aqueous phase.*

*Sample 1630-PC, from the vent spire (Fig. 2), was subsampled and sealed immediately upon arrival onboard. The distribution of the volatile hydrocar bons is shown in the GC trace (Fig. 15); they range from methane to n-decane, a typical condensate petroleum. It should be noted that homologs </j-C5 are depleted relative to typical reservoir condensates due to volatile loss during transit of the submersible through the thermocline and during recovery. Minor amounts of benzene and toluene are also present, as is the case for vent fluids analyzed from Guaymas Basin (Simoneit et al. unpub. data). These aromatic hydrocarbons are derived from high-*





FiG. 12. Gas chromatograms for total extracts (n-heptane) of clasts (oil veins) from core 1630-AC1: (a) Section 6; (b) Section 7, "shale" piece; (c) Section 7, other clasts; (d) Section 9; (e) Section 9. Peak identificatio FIG. 12. Gas chromatograms for total extracts (n-heptane) of clasts (oil veins) from core 1630-ACt: (a) Section 6; (b) Section 7, "shale" piece; (c) Section 7, other clasts; (d) Section 8; (e) Section 9. Peak identificatio







s. 14, Setecter mass rragmentograms (m/2 +cosas xem mantocr) row we procedure interest (c,i) m/z 192 (4 = methylphemathrenes); (d,i) m/z 206<br>m/z 178 (1 = phemanthrene, 2 = anthracene); (b,h) m/z 190 (3 = methylcnephemanth zolinipation, pyrone and persention.<br>Fig. 14. Selected mass fragmentograms (*m/z versus* scan number) for the phenanthrene series of sample 1630-AC1, 7 (a-f) and sample 1630-PC (g-l): (a.g)<br>Fig. 14. Selected mass fragment Fig. 14. Selected mass fragmentograms (*m/z versus* scan number) for the phenanthrene series of sample 1630-AC1, 7 (a-f) and sample 1630-PC (g-l): (a,g)<br>  $m/z$  178 (1 = phenanthrene, 2 = anthracene); (b,h)  $m/z$  190 (3 = me





*MPt=methylphenanihrene index;DNR=dimethylnaphihalene ratio*

*temperature catagenetic processes (Whelan el al. 1988). The ratios of the iso- to the normal alkanes (C, and C5) are in the range of 3.3 to 1.0 and can be interpreted as immature to mature (Thompson 1979, Whelan et al. 1988). Solvent extracts of the bulk sample 1630-PC exhibit identical GC traces for interior (in contrast with vent fluid) and exterior subsamples.GC-MS fingerprinting (e.g.. Fig. 13b) indi cates that samples 1630-PC and 1630-ACI have different compositions, partly as a result of losses of the normal and isoprenoidal alkanes from the lat ter. The dominant n-alkancs range from <C, to C,0, with major amounts of isoprcnoid hydrocar bons* ( $C_{10}$  *to*  $C_{20}$ , *no*  $C_{12}$  *or*  $C_{17}$ ; *Fig.* 13*b*). Again, *polycyclic biomarkcrs arc not detectable, which, cou pled with the resolved hydrocarbon range to C20, is consistent with a condensate petroleum composition.*

*The major PAHs in sample 1630-PC consist of phenanthrene, anthracene, fluoranthcne, pyrenc, benzanthracene, chrysene, benzfluoranthene, benzpyrenes, and perylene, with lesser amounts of thio-, oxo- and alkyl-PAH. The PAH distribution and the presence of peri-condensed alicyclic PAH (e.g., mcthylenephenanthrenc, Fig. 14h) also indicate a*

*high-temperature component reflecting hydrother mal activity. However, the distribution of thesecom pounds in 1630-PC is not identical to that sample 1630-ACI,7 (cf. Fig. 14h-l versus Fig. 14b-f) and is due to unknown influences.*

*The dimethylnaphthalene (DNR) and methylphenanthrene (MP!) indices for the chimney sam ple (Table 2) fall in the immature and oil or conden sate rangewindow (Radke & Welte 1983). This is also consistent with a short migration distance from source to seabed, and the alkane distribution sug gests no biodegradation occurred en route.*

*In summary, the petroleum in dive area 1630con sists of immature condensates containing hydrocar bons* from methane to about *n*-C<sub>20</sub>, with admixed *PAH components from high-temperature (> - SOO^C) hydrothermal alteration oforganic mat ter. In the mound/chimney system (Fig. 2) this con densate is present as unaltered oil (i.e., nonbiodegraded, Simoneit 1985a), even at the exterior surface of the chimney/mound matrix(sample 1630- PC). Thisindicates that active petroleum migration exceeds microbial degradation. Condensate migra tion seems to be from shallow depth (i.e., few meters sub-bottom, consisting of the early petroleumgeneration products moving over a short distance) and by bulk-phase movement, completely saturat ing the young mound/vent system. Based on visual observation, most of these condensates are lost in such areas to the seawater and rise as plumes (Merewether et al. 1985, Simoneit et al. in prep.). On the other hand, the petroleum in the sediment surrounding the mound/vent system (core 1630- ACI) is a completely biodegraded condensate, where only the UCM and the PAH remain. Makingan oiloil correlation, i.e., matching biomarker and PAH compositions, this material is the same as the unaltered petroleum in the mound/vent area. Most of the petroleum in the core is present as vein fill-*





*ings in silica clasts (Fig. 4). Because the wallrock is porous silica and not carbonate, it is concluded that the oil filled fractures by bulk-phase migration and was biodegraded during deposition in situ. Addi tional and variable degradation and loss of volatiles occurred after recovery. Carbonate precipitation of the respiratory carbon dioxide from microbial petroleum degradation was not obvious in the core.*

#### *Conclusions*

*Studies of the composition of interstitial waters and petroleums obtained from short ALVIN push cores in an area of upwelling fluids in Guaymas Basin have revealed the following:*

*1. Water samples taken in 1988 from hydrothermal vents indicate a hydrothermal fluid end-member chemistry very similar to those observed in 1982and 1985, thus indicating a long-term stability of the hydrothermal system active in this area.*

*2. In regions characterized by shimmering waters overlying the sediments, "diffuse" flow occurs through the high-porosity sediments as well as through small holes. Core 1617-ACI, taken in an area of diffuse flow, contains porewaters with a sub stantial hydrothermal component (low Mg2\*, SO.,2 , high Ca2+, Li\*, K+, CI"), which becomes constant below a depth of 4 cm into the sediment.*

*3. In areas characterized by the formation of Beg gialoa bacterial mats, upward-moving porewaters have a much more diluted hydrothermal component (e.g., cores 1619-AC2, I629-AC2). A strong deple tion of sulfate in the upper 10 cm of the sediment leads* to a large flux of sulfide to the sediment-water *interface. This sulfide is then in turn consumed by the Beggialoa. We postulate that bacterial sulfate reduction is enhanced both as a result of higher tem peratures in these sediments, and increased nutrients in the rising porewaters.*

*4. In layers enriched in diatoms, cementation ofsedi ments by silica precipitation can lead to the block age of upward-flowing hydrothermal fluids. This results in channelling of migrating hydrothermal fluids to fractures associated with faults, and thereby focuses the location of hydrothermal deposits on the seafloor. Core 1630-ACI is representative of such cementation.*

*5. A comparison of interstitial-water compositional changes in cores 1617-ACI, 16I9-AC2, 1629-AC2, 1630-ACI and 1984-ACI indicates that they represent mixtures of seawater and hydrothermal fluids; in situ chemical reactions have caused over prints on these changes, so that the data cannot be used to distinguish whether the hydrothermal endmember is typically associated with a deep-seated hydrothermal system or with a sill-induced hydrothermal system.*

*6. In dive area 1630 large amounts of petroleum (both oil and gas) are associated both with hydrother* *mal spires/mounds (1630-PC) and nearby sediments (1630-ACI). In the sediments, the petroleum occupies mainly fractures in silica-cemented diatomaccous mudstonc clasts. Gcochemical analyses of the petroleum at both sites have shown that it con sists of immature condensates with a hydrothermal PAH component. In core 1630-ACI the petroleum is biodegraded.*

*7. Petroleum has and is actively migrating in bulk* as a condensate  $(CH_4$  to  $\sim n-C_{20}$ , the early *petroleum-generation products) over relatively short distances (less than a few meters). The steep ther mal gradient, reaching - 100°C at 1 m sub-bottom, provides the drive for this migration; most of the oil emanates to the seawater rising as plumes.*

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